

**XXII INTERNATIONAL SYMPOSIUM
„ADVANCES IN THE CHEMISTRY OF
HETEROORGANIC COMPOUNDS”**



**Centre of Molecular
and Macromolecular Studies
Polish Academy of Sciences**

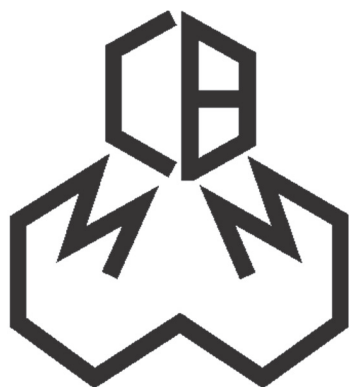


**Section
of Heteroorganic Chemistry
Polish Chemical Society**

**ŁÓDŹ
November 22, 2019**

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ORGANIZED BY



Section of Heteroorganic Chemistry
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in cooperation with

Faculty of Chemistry
University of Łódź

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Jan Długosz University
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Łódź Branch
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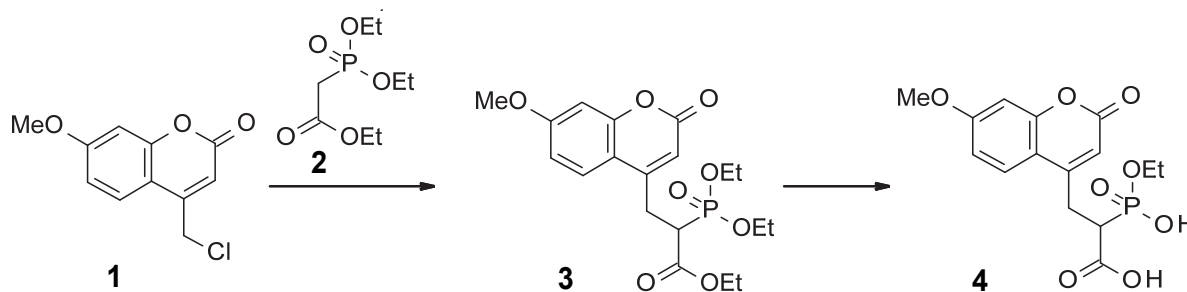
Synthesis of Coumarin-4-ylethyl Phosphonic Acid Derivatives

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Naturally occurring coumarin umbelliferone (7-hydroxycoumarin) and its *O*-methyl ether (herniaryn, 7-methoxycoumarin) exhibit different types of biological activity. Modifications of coumarins made it possible to obtain biologically active molecules, which is important in the drug design.[1, 2] In order to expand the diversity of bearing phosphonic acid moiety coumarins, the possibility for the synthesis of homologues previously obtained 4-phosphonomethylcoumarins with antiviral activity [3] was studied.



Scheme 1. Synthesis of coumarin-4-ylethyl phosphonic acid derivatives.

We investigated the reaction of 7-methoxy-4-chloromethylcoumarin (1) and ethyl (diethoxyphosphoryl) acetate (2) in various solvents (1,4-dioxane, dimethyl ether ethylene glycol, dimethylformamide, 1-methylpyrrolidone, 1,3-dimethyltetrahydropyrimidin-2-one) in the presence of various bases (potassium *tert*-butoxide, sodium hydride, lithium or sodium hexamethyldisilazide), which led us to the synthesis of ethyl 2-(diethoxyphosphoryl)-3-(7-methoxy-2-oxo-2H-chromen-4-yl) propanoate (3). It was found that the ratio of 4-chloromethylcoumarin / ethyl (diethoxyphosphoryl) acetate / sodium hydride 1: 2: 2 using dimethylformamide as the solvent was optimal. The reaction proceeds at a temperature of 25-35°C for 7-14 days.

Alkaline hydrolysis of ester 3 with lithium hydroxide excess affords new 2-(ethoxyphosphoryl)-3-(7-methoxy-2-oxo-2H-chromen-4-yl) propionic acid (4).

The structure and purity of compounds 3 and 4 were confirmed by spectroscopic studies: ¹H NMR, ¹³C and ³¹P and LCMS.

References

- [1] S. C. Demmer *et al.*, *Chem. Rev.*, **2011**, *111*, 7981.
[2] F. Borges *et al.*, *Cur. Med. Chem.*, **2005**, *12*, 887.
[3] K. M. Kondratyuk *et al.*, *Chem. Nat. Compd.*, **2019**, *55*, 632.